

# Reduction of 4-(Haloalkyl)azetid-2-ones with $\text{LiAlH}_4$ as a Powerful Method for the Synthesis of Stereodefined Aziridines and Azetidines

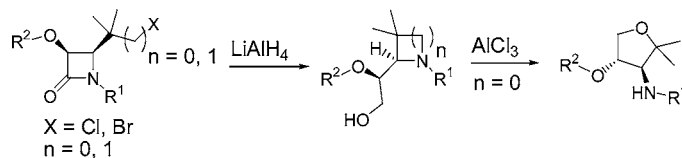
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## ABSTRACT



A new synthesis of stereodefined aziridines and azetidines, starting from 4-(1- or 2-haloalkyl)azetid-2-ones, is described. Treatment of the latter compounds with  $\text{LiAlH}_4$  gave 1,2-fission of the  $\beta$ -lactam, followed by an intramolecular nucleophilic substitution of the halogen, giving rise to the formation of 2-(1-alkoxy-2-hydroxyethyl)aziridines in the case of 4-(1-haloalkyl)azetid-2-ones and of 2-(1-alkoxy-2-hydroxyethyl)azetidines in the case of 4-(2-haloalkyl)azetid-2-ones. The resulting 2-(1-alkoxy-2-hydroxyethyl)aziridines were transformed into the corresponding trans-3,4-substituted oxolanes via an intramolecular nucleophilic ring opening, triggered by  $\text{AlCl}_3$ .

The use of  $\text{LiAlH}_4$  as a reductive agent of azetid-2-ones has been shown not to be predictive, as both direct reduction of the carbonyl function<sup>1–4</sup> (leading toward azetidines) and 1,2-fission of the  $\beta$ -lactam core<sup>4–8</sup> (leading toward  $\gamma$ -amino alcohols) may occur. In addition, the former reaction pre-

dominantly occurs with N-unsubstituted  $\beta$ -lactams, whereas the 1,2-fission is mostly described for azetid-2-ones bearing an electron-withdrawing group (Boc, sulfonyl) or an aryl at the nitrogen.

4-(1- or 2-Haloalkyl)azetid-2-ones consist of an interesting class of azaheterocyclic compounds because the presence of ring strain and a halogen in these compounds makes them attractive for further reactions. In addition, these  $\beta$ -lactams are now readily available by performing a Staudinger reaction between  $\alpha$ -chlorinated imines and different ketenes (prepared in situ from the corresponding acid chlorides and triethylamine). Investigating the reactivity of these compounds toward  $\text{LiAlH}_4$  seems interesting, as treatment of 4-(1-haloalkyl)azetid-2-ones **1** with  $\text{LiAlH}_4$  may give rise to the formation of 2-(1-haloalkyl)azetidines **3** or 2-(1-alkoxy-2-hydroxyethyl)aziridines **2** depending on whether reduction of the  $\beta$ -lactam carbonyl or 1,2-fission of the  $\beta$ -lactam amide

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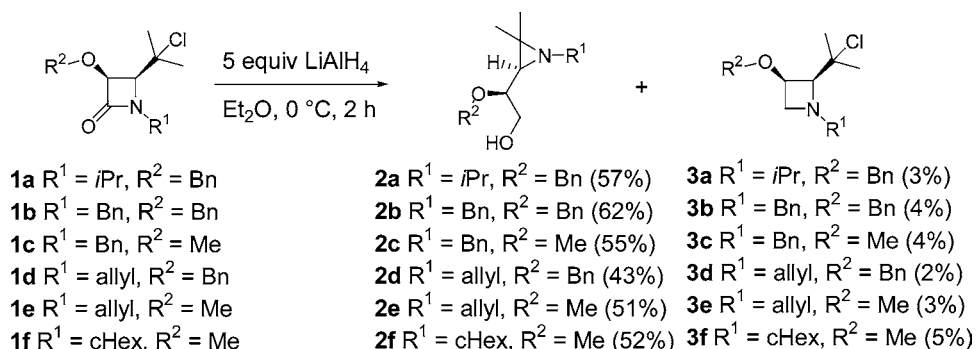
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**Scheme 1.** Stereoselective Synthesis of 2-Hydroxyaziridines **2**



bond occurs. The former compounds are very attractive starting materials to perform rearrangements toward 3-halo-pyrrolidines,<sup>9–12</sup> whereas the latter serve as excellent building blocks in a variety of syntheses.<sup>13–19</sup> Treatment of 4-(2-haloalkyl)azetidines **11** with LiAlH<sub>4</sub> may afford in the same way 2-(2-haloalkyl)azetidines **12** or 2-(1-alkoxy-2-hydroxyethyl)azetidines **13**. In general, azetidines constitute an interesting class of four-membered heterocycles, both from synthetic and from medicinal viewpoints.<sup>20–27</sup>

The present research will focus on the reactivity of LiAlH<sub>4</sub> toward 4-(1- or 2-haloalkyl)azetidines **1** and **11** to

develop new pathways for synthetically interesting aziridines and azetidines.

To investigate the reaction of LiAlH<sub>4</sub> with 4-(1-haloalkyl)-azetidines **1**, synthesized by [2+2] cyclocondensation of the corresponding halogenated imines and ketenes (formed in situ from the corresponding acid chlorides by treatment with triethylamine) in benzene,<sup>28,29</sup> the latter compounds were added to a mixture of 5 molar equiv of LiAlH<sub>4</sub> in diethyl ether at 0 °C. After 2 h, a reaction mixture of mainly 2-(1-alkoxy-2-hydroxyethyl)aziridines **2** and little 2-(1-haloalkyl)-azetidines **3** (2–5%) was obtained. Separation of these two compounds by flash chromatography proved to be very easy, as both compounds seriously differ in polarity. After purification, the 2-(1-alkoxy-2-hydroxyethyl)aziridines **2** were isolated in good to high yields (Scheme 1). It has to be noted that the order of addition turned out to be very important during this reduction. Addition of LiAlH<sub>4</sub> to a mixture of 4-(1-haloalkyl)azetidines **1** in diethyl ether gave a very substantial decrease of yields, and many side products were formed, indicated by the complex reaction mixtures obtained after workup.

As can be seen in Scheme 1, ring transformation of 4-(1-chloroalkyl)azetidines **1** toward aziridines **2** proceeds with retention of stereochemistry. The 1,2-fission of the azetidines **1** explains the obtained syn stereochemistry observed in aziridines **2**. The proposed mechanism proceeds via an initial reduction of the amide functionality into an hemiaminal **4**. Coordination of lithium to nitrogen triggers ring opening of azetidinium salt **5**, leading to the formation of the intermediate aminoaldehyde complex **6**. Reduction of the formyl group and nucleophilic substitution of the chlorine by nitrogen leads to the formation of 2-(1-alkoxy-2-hydroxyethyl)aziridines **2** (Scheme 2).<sup>4</sup> The obtained stereochemistry of aziridines **2** was proven by the observation that the vicinal coupling constants (*J* = 8.6 Hz; <sup>1</sup>H NMR; CDCl<sub>3</sub>) between the NCH of the aziridine ring and the R<sup>2</sup>OCH in all cases were similar to coupling constants found in the literature for aziridines with analogous stereochemistry.<sup>29</sup>

The formation of small amounts of azetidines **3** could be due to two reasons. Alane, formed after initial reaction of

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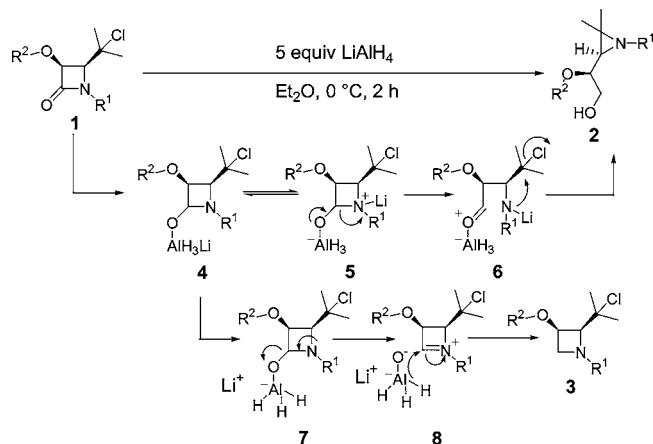
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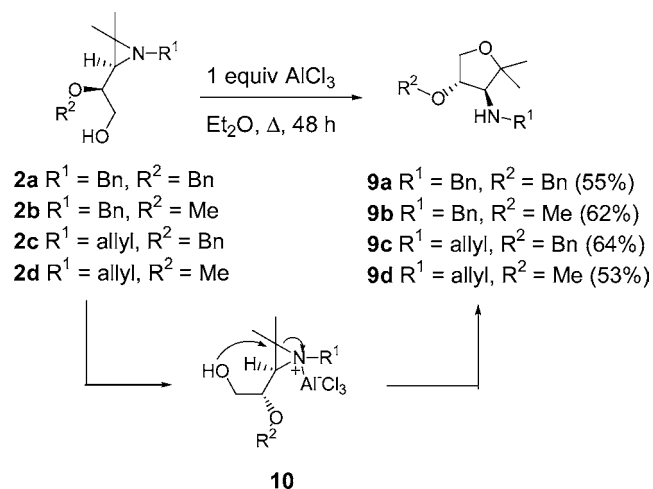
**Scheme 2.** Reduction Reaction Mixtures of 2-(1-Haloalkyl)azetidines **3**



LiAlH<sub>4</sub> with azetidin-2-ones **1**, could be responsible for reduction of the latter compounds into azetidines **3**.<sup>4</sup> On the other hand, formation of an azetidinium ion **8** can also give rise to the formation of azetidines **3** (Scheme 2). The observation that the use of more equivalents of LiAlH<sub>4</sub> did not result in a diminishing amount of azetidines **3** proved that alane was not responsible for the formation of 2-(1-haloalkyl)azetidines **3**. Reduction of immediately formed azetidinium salts **8** is more suitable to explain the observed results (Scheme 2).

Aziridines have been shown to serve as excellent building blocks in organic synthesis.<sup>13–19</sup> Also, aziridines **2** proved to be very useful intermediates toward the synthesis of highly substituted oxolanes. Therefore, 2-(1-alkoxy-2-hydroxyethyl)aziridines **2** were treated with AlCl<sub>3</sub> in ether under reflux for 48 h, affording *trans*-tetrahydrofurans **9** in good isolated yields (Scheme 3). The stereochemistry was clearly shown

**Scheme 3.** Stereoselective Synthesis of 3,4-*trans*-Tetrahydrofurans **9**

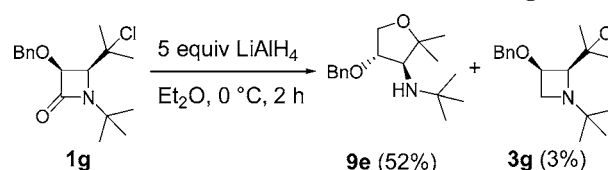


to be *trans*, as coupling constants of 4 Hz were observed

between the two protons at C3 and C4 (<sup>1</sup>H NMR; CDCl<sub>3</sub>). The reaction mechanism is explained by an initial complexation of AlCl<sub>3</sub> with nitrogen. In this way, the activation of the aziridine ring renders a more facile ring opening after intramolecular nucleophilic attack of the hydroxyl function, leading toward oxolanes **9**. Recently, unusual dideoxy-nucleosides received increasing interest because of their antiviral activity.<sup>30,31</sup> More specifically, *trans*-3-hydroxy-4-uracyloxolane proved to show HCMV inhibitory activity.<sup>31</sup>

It has to be noted that, when 3-benzyloxy-1-*tert*-butyl-4-(1-chloro-1-methylethyl)azetidin-2-one **1g** was treated with 5 molar equiv of LiAlH<sub>4</sub> at 0 °C for 2 h, no 2-(hydroxyethyl)aziridine **2** could be obtained, as only the corresponding oxolane **9e** was isolated, next to small amounts of azetidine **3g** (Scheme 4). Apparently, in this case, the *tert*-butyl group

**Scheme 4.** Reduction of Azetidin-2-one **1g**



exerts too much strain onto the intermediate 2-(hydroxyethyl)aziridine to be stable under these reaction conditions, and intramolecular attack of the hydroxyl function leads toward the formation of oxolane **9e**.

Treatment of 4-(2-bromoalkyl)azetidin-2-ones **11**, prepared by [2+2] cyclocondensation of the corresponding *N*-(3-bromo-2,2-dimethylpropylidene)alkylamines and ketenes (formed in situ from the corresponding acid chlorides by treatment with triethylamine) in benzene<sup>29</sup> with 5 molar equiv of LiAlH<sub>4</sub>, gave analogous results as with 4-(1-chloroalkyl)azetidin-2-ones **1**. However, because of the extended carbon chain, 1,2-fission of the starting material followed by a nucleophilic substitution of bromide led toward the formation of 2-(1-alkoxy-2-hydroxyethyl)azetidines **12** next to small amounts of *cis*-4-(2-bromoalkyl)azetidines **13** (1–5%), obtained by reduction of the carbon–oxygen double bond of 4-(2-bromoalkyl)azetidin-2-ones **11** (Scheme 5). Again, the obtained *syn* stereochemistry of azetidines **12** was proven by the observation that coupling constants (*J* = 9.4 Hz; CDCl<sub>3</sub>) between the NCH of the azetidine ring and the R<sup>2</sup>-OCH units were similar to coupling constants found in the literature for azetidines with analogous stereochemistry.<sup>29</sup>

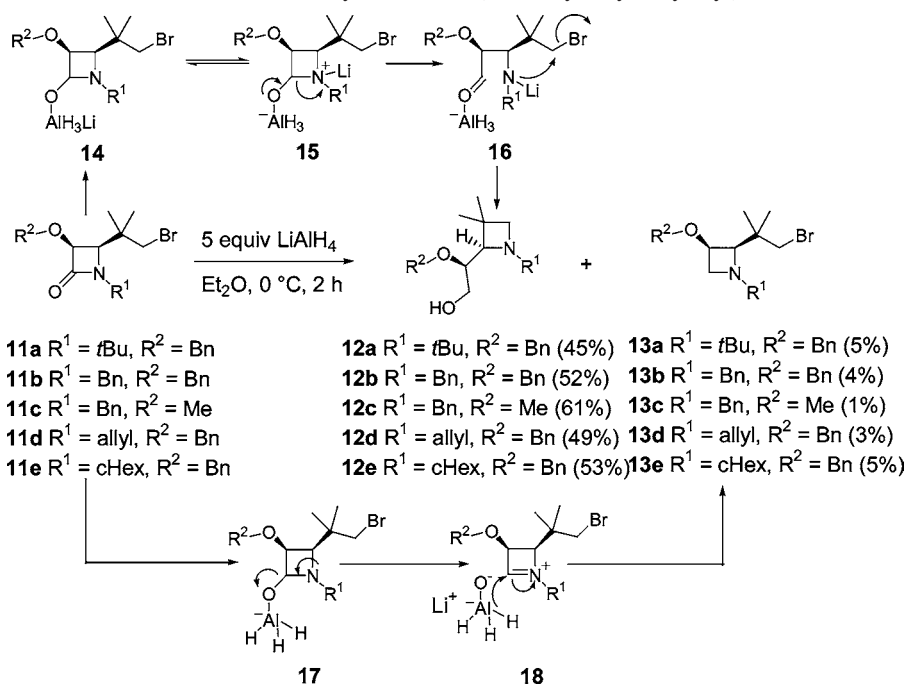
Attempts to perform ring enlargements of 2-(hydroxyethyl)azetidines **12** toward pyranes were not successful.

In conclusion, 4-(haloalkyl)azetidin-2-ones were proven to be useful synthons in the synthesis of new functionalized aziridines and azetidines. Treatment of 4-(haloalkyl)azetidin-2-ones with LiAlH<sub>4</sub> gave 1,2-fission of the β-lactam core, followed by a nucleophilic substitution of the halogen, giving

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**Scheme 5.** Stereoselective Synthesis of 2-(1-Alkoxy-2-hydroxyethyl)azetidines **12**



rise to the formation of stereodefined 2-(1-alkoxy-2-hydroxyethyl)aziridines in the case of 4-(1-haloalkyl)azetid-2-ones and of 2-(1-alkoxy-2-hydroxyethyl)azetidines in the case of 4-(2-haloalkyl)azetid-2-ones. In all cases, very small amounts of azetidines, originating from the reduction of the carbonyl function on the  $\beta$ -lactam core, were formed as side products. Furthermore, 2-(1-alkoxy-2-hydroxyethyl)aziridines could be transformed into the corresponding trans-3,4-substituted oxolanes via an intramolecular nucleophilic ring opening, triggered by AlCl<sub>3</sub>.

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**Supporting Information Available:** General information and all spectroscopic data. This material is available free of charge via the Internet at <http://pubs.acs.org>.

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